



Nanotechnologies used in NANO CI

Nanotechnology-based implantable and interfaceable devices

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Herbert Keppner

Partnership



Bar-Ilan University

MED⁹EL



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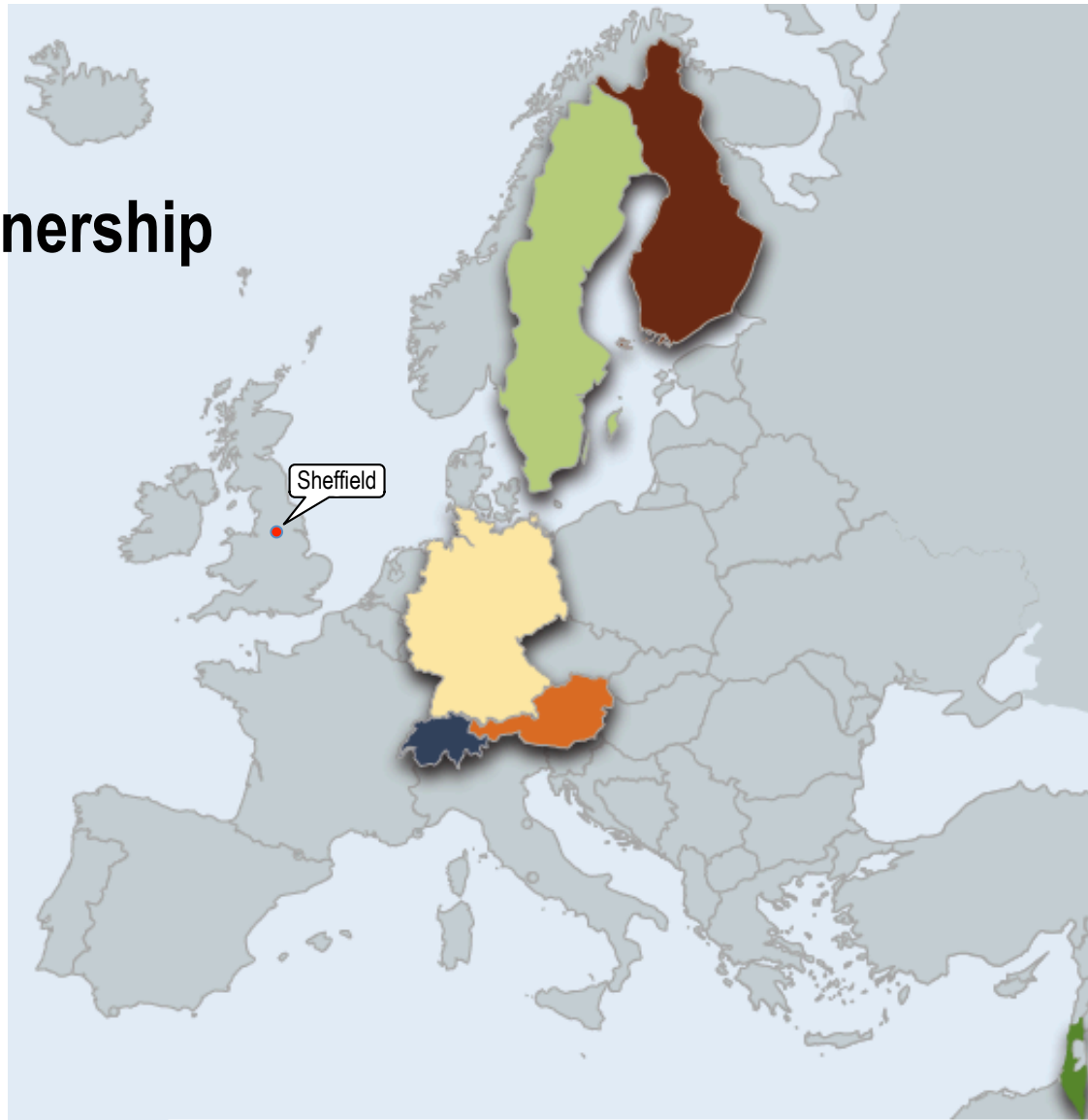











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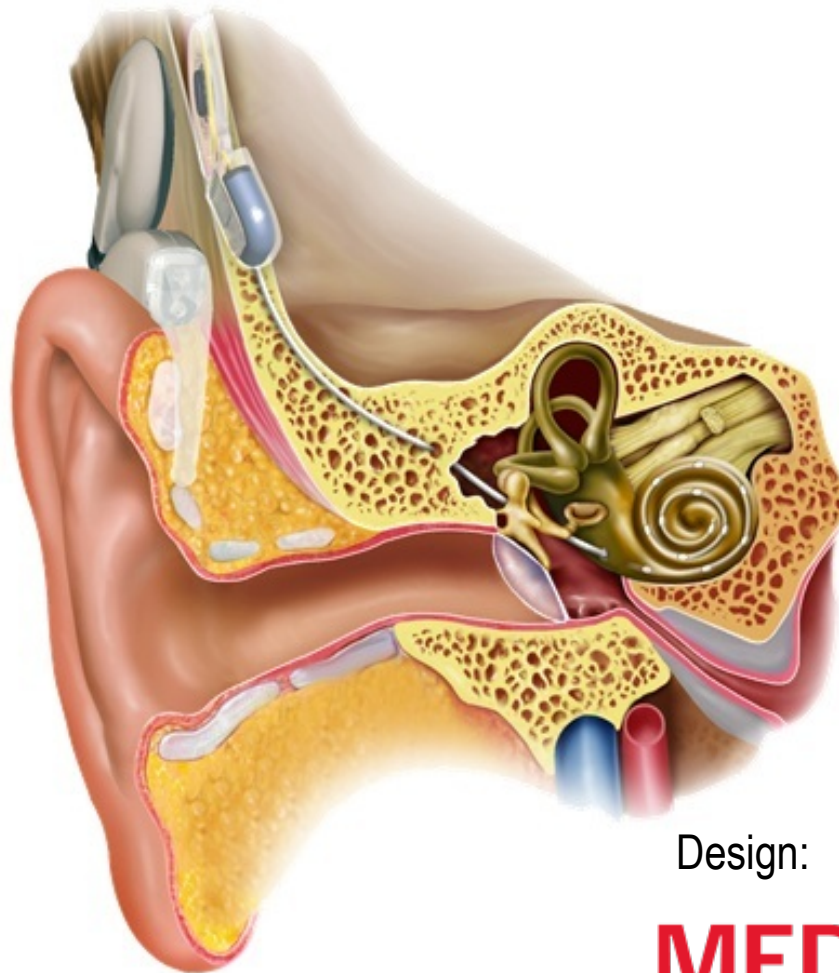
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2	UT Tübingen, DE	
3	UU Upsala, SE	
4	UTA Tampere, FI	
5	HES-SO Délemont, CH	
6	BIU Tel Aviv, IL	
7	EMC Tübingen, DE	
8	MED-EL Innsbruck, AT	
9	SCIPROM Lausanne, CH	

Outline

- Summary of the challenges
- Selected technologies
 - drug dispensing:** SOLID on Liquid deposition.
 - surface conditioning:** Deposition of nanoparticles by sonochemistry
- Conclusion and outlook.

Summary of the challenges

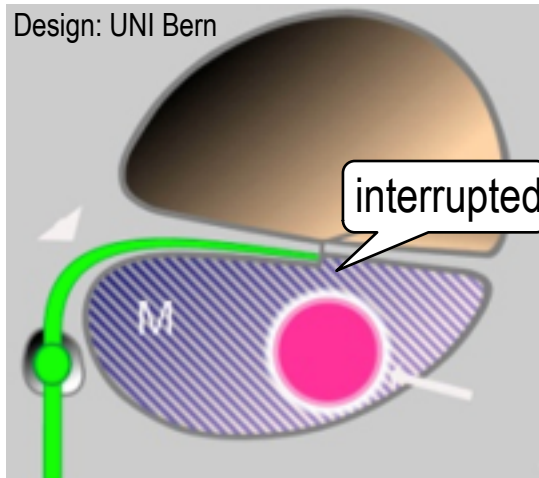
Development of a cochlear implant



Design:

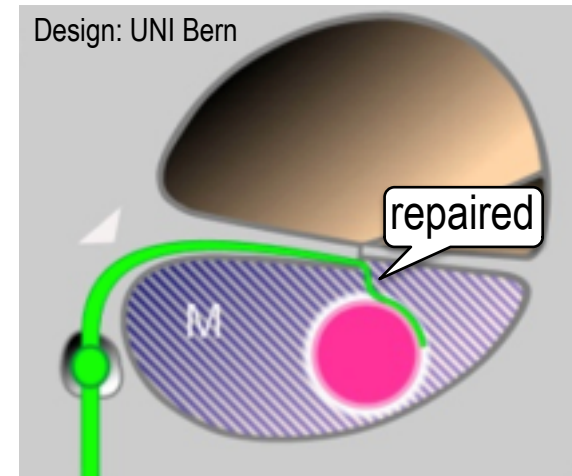
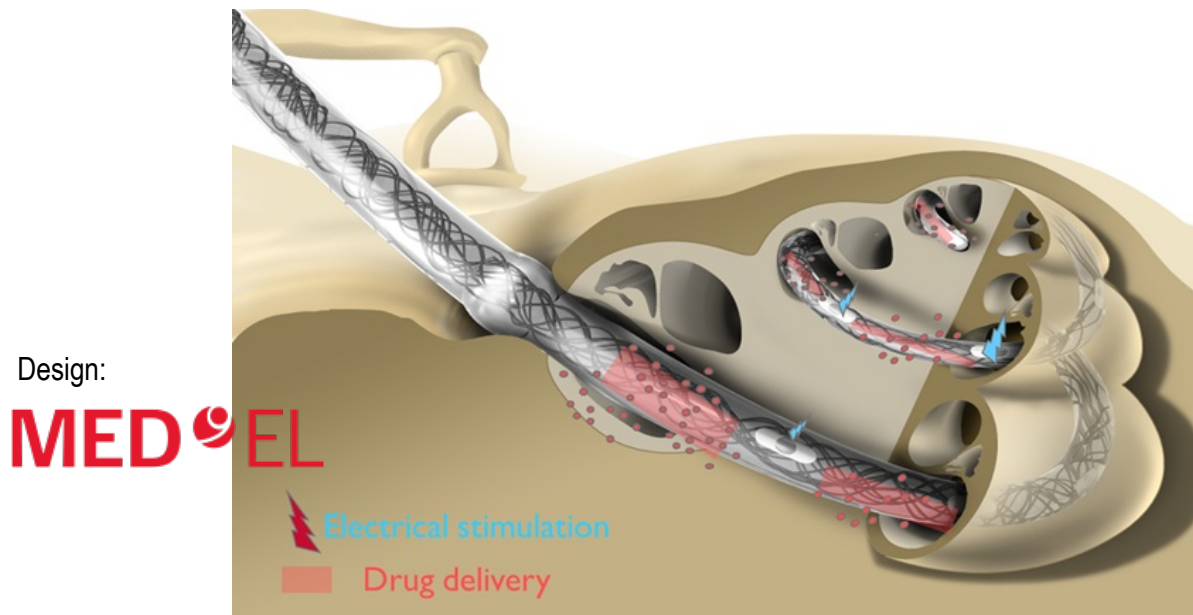
MED  **EL**

Summary of the challenges

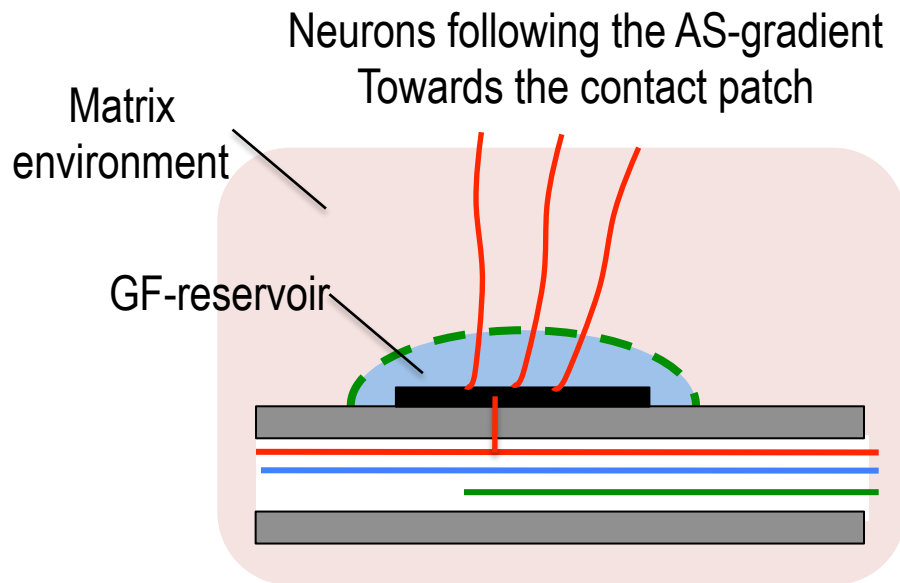
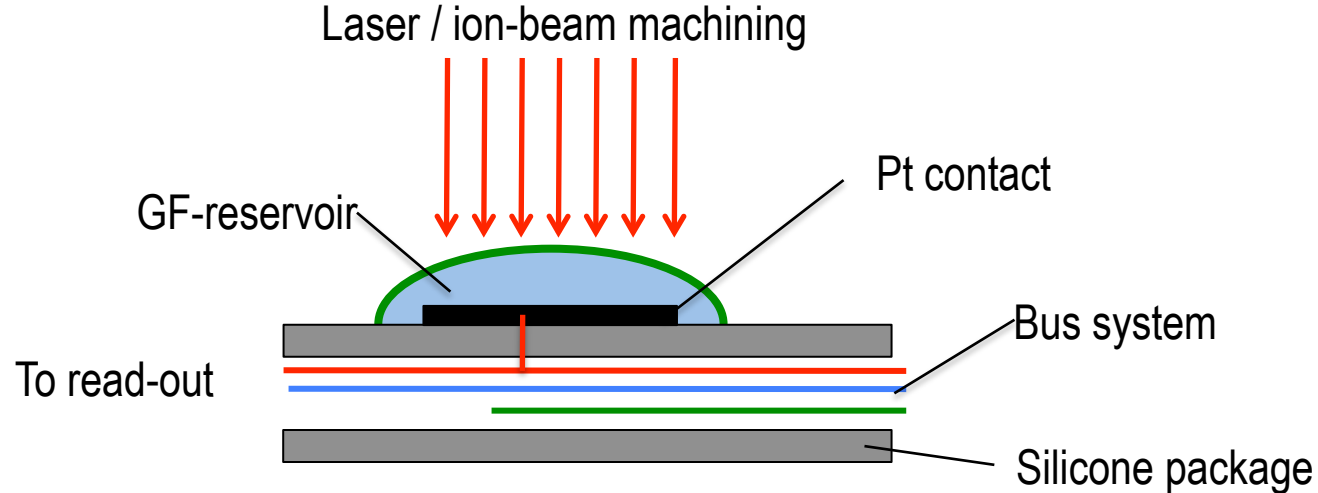


Essential elements for neuronal interfacing:

- Attraction of neuron-growth towards electrodes via selectively operating delivery of growth factors.
- Creation of a neuron-electrode junction allowing signal transmission.



Growth Factor (GF) emitting device (vision)



Technological Issue:
how to package the liquid?
Solid on liquid deposition wanted

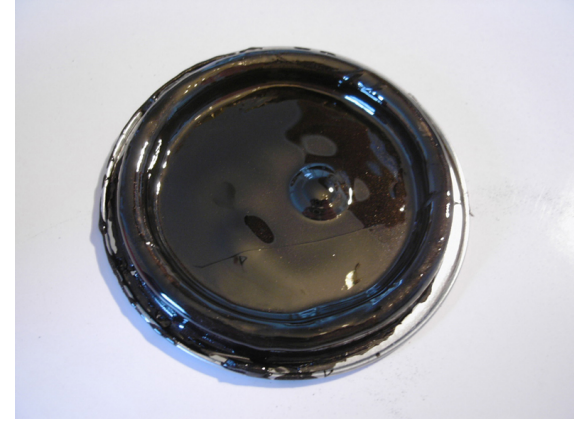
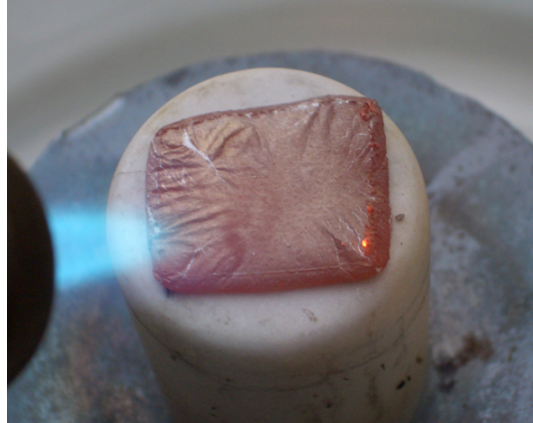
Solid on liquid deposition

Finding a solution: short walk in the garden:



Solid on liquid?

More and others



Natural SOLID systems seems to be either thermally, mechanically or chemically **unstable**

First “archaic” approach for obtaining better stability

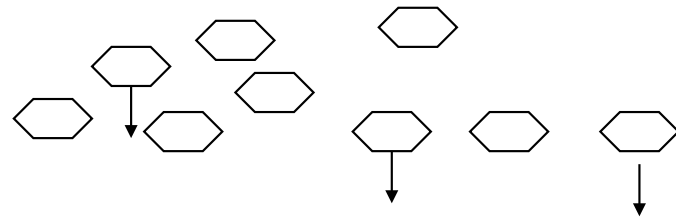
Mercury has a high density; paint spray come from the gas phase and polymerises on the surface of a Mercury droplet.



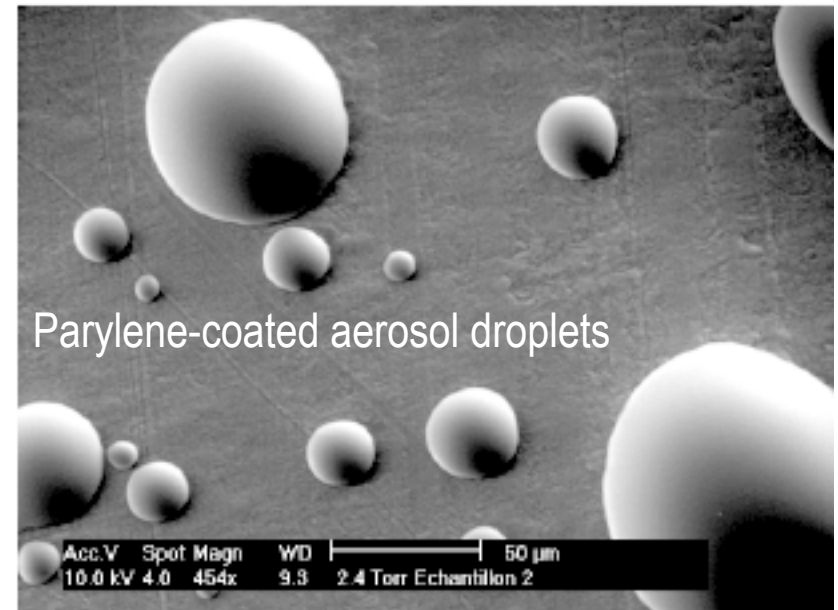
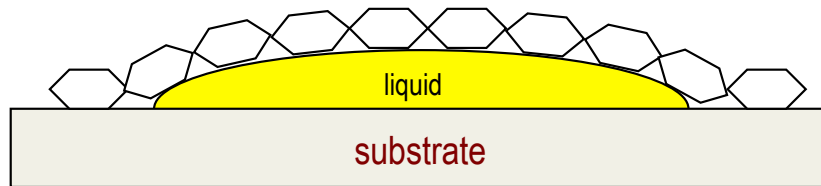
SOLID (Solid on liquid deposition): Coating of liquids and hermetical sealing using CVD Parylene.

H. Keppner, M. Benkhaira, Patent WO/2006/06395, Dec. 2004.

J. Charmet et al. Thin Solid Films 518 (2010) 5061–5065



Monolayer of Parylène

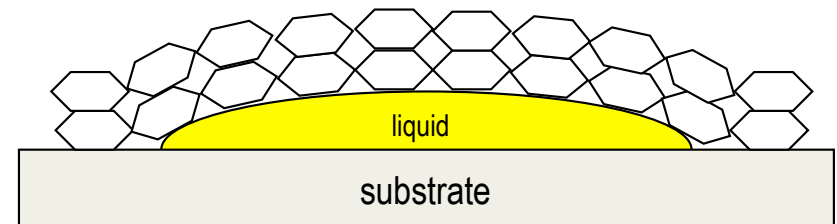


Parylene-coated aerosol droplets



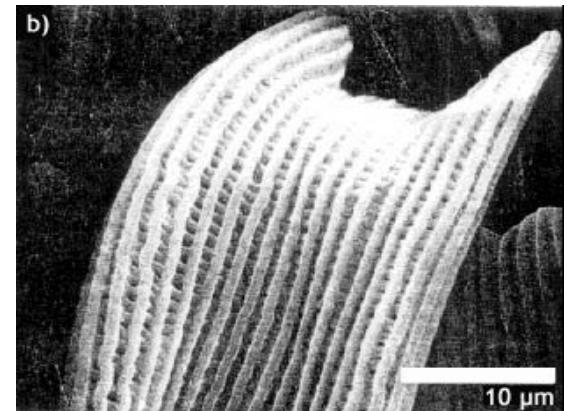
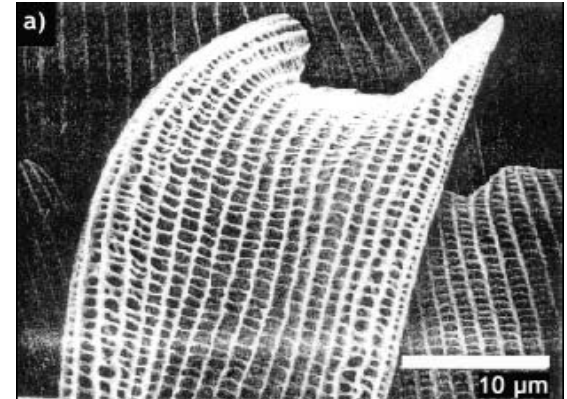
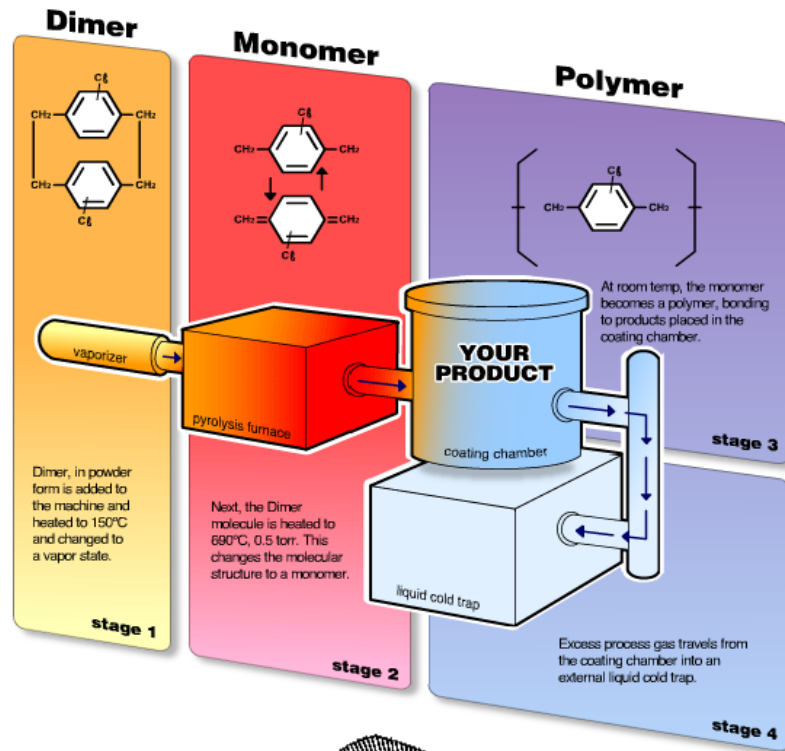
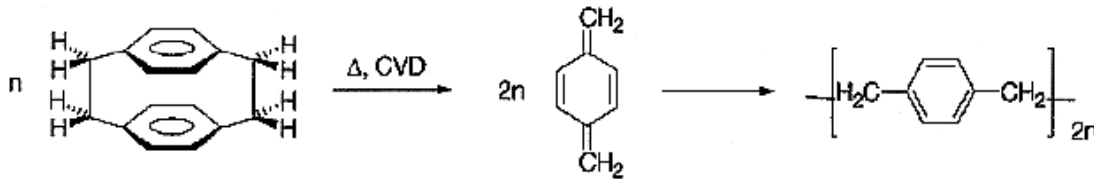
Parylene sealed silicon-oil drop attached at an optical fibre

Later: mechanically stable coating

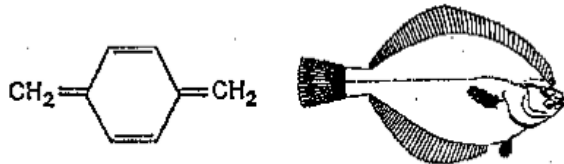


LPCVD deposition of Parylene (Poly-Chloro-Para Xylylene) :

W.F. Gorham, J. Poly, Sci., Part A-1: Polym. Chem. 4 (1966) 3027



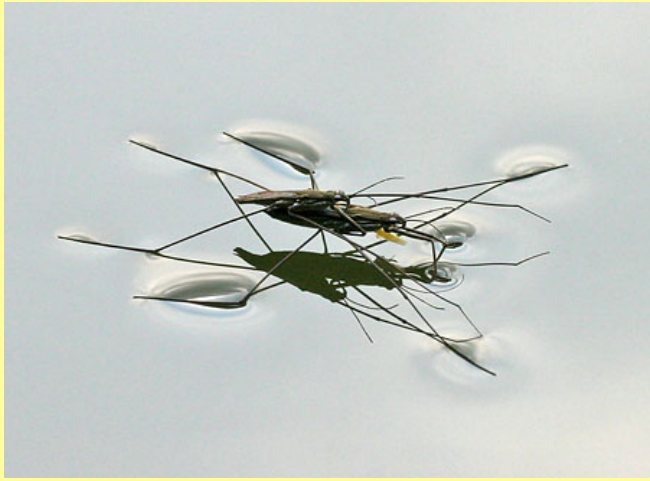
SEM micrograph from a Butterfly wing a) uncoated, b) Parylene coated
D.W. Grattan; *Can. Chem. News* **1989**, 25



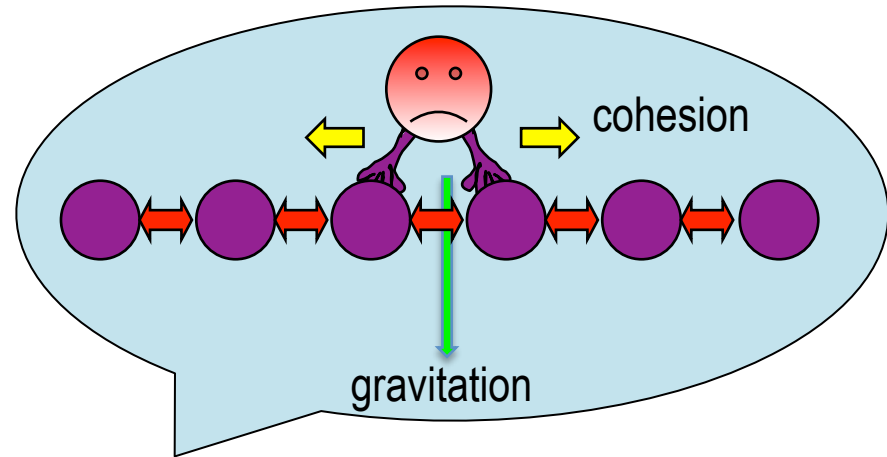
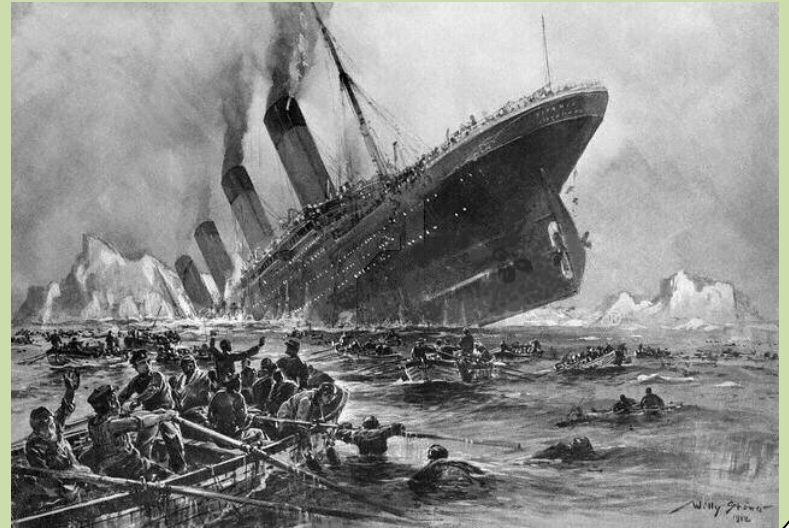
Parylene layers are highly biocompatible

Why SOLID-technology may be considered as nanotechnology:

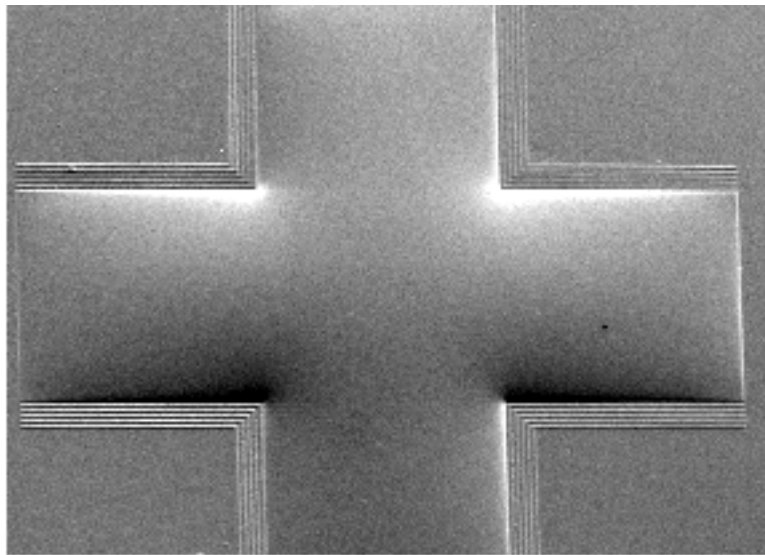
Large, high-density solid clusters tend to sink as soon as Archimedes' principle is not fulfilled



Smaller solid clusters have to overcome the surface energy and can swim.

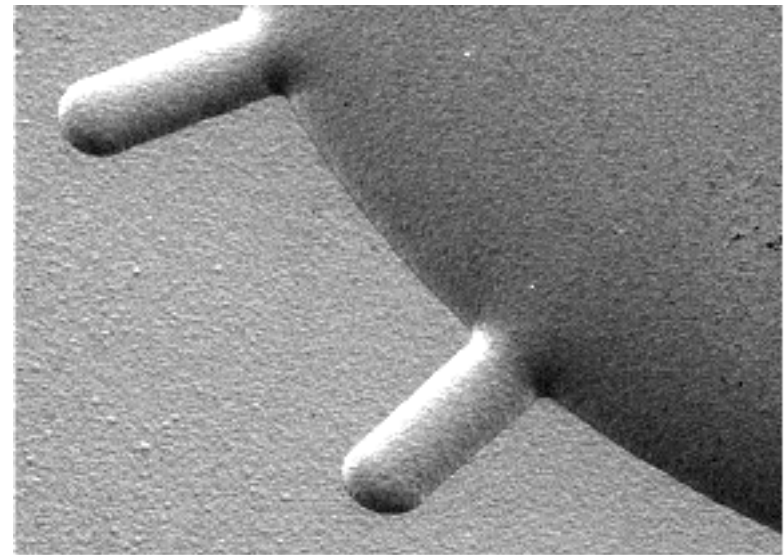


At molecule level, atoms or molecules are much more affected by cohesion forces rather than to gravitation.



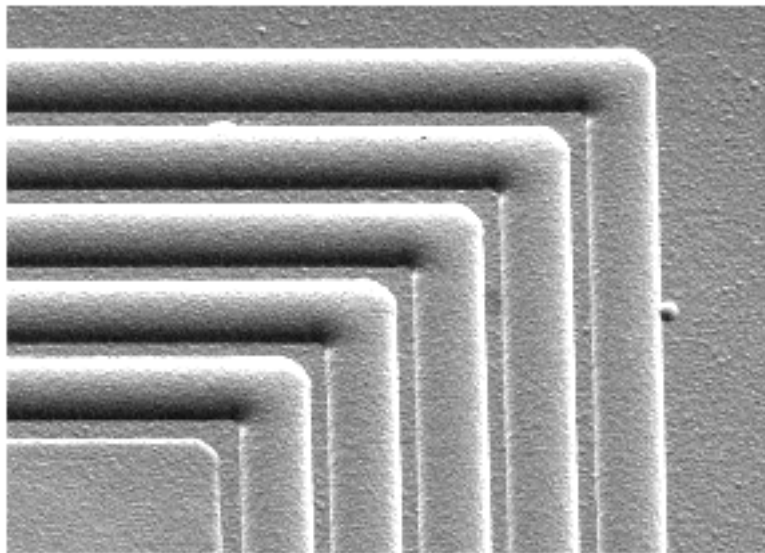
5kV [SE] WD=14mm · Solid_cross_1

300 μm



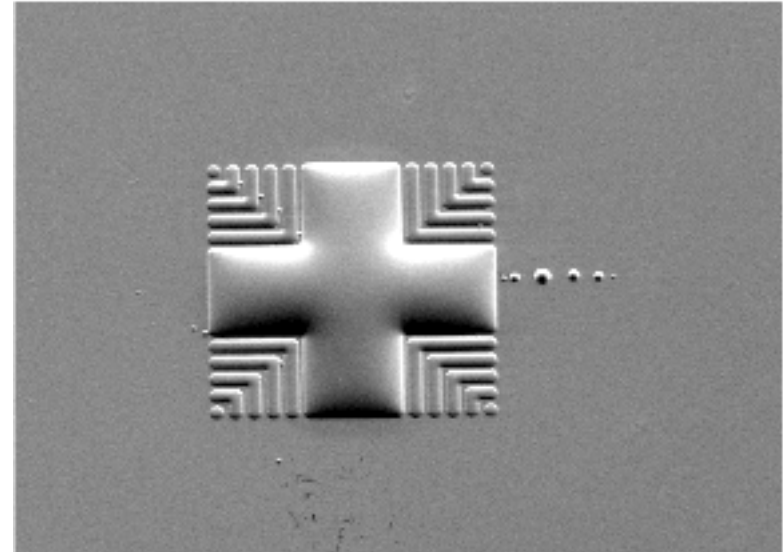
5kV [SE] WD=14mm · Solid_circle_4

10 μm



5kV [SE] WD=14mm · Solid_canal_1

20 μm



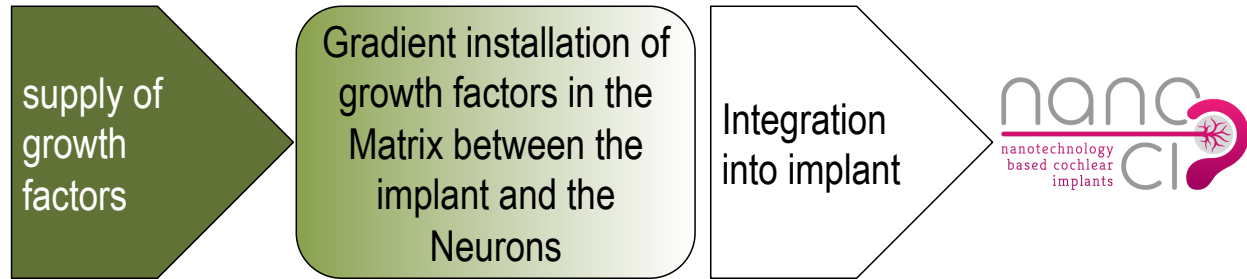
5kV [SE] WD=14mm · Solid_cross_2

50 μm

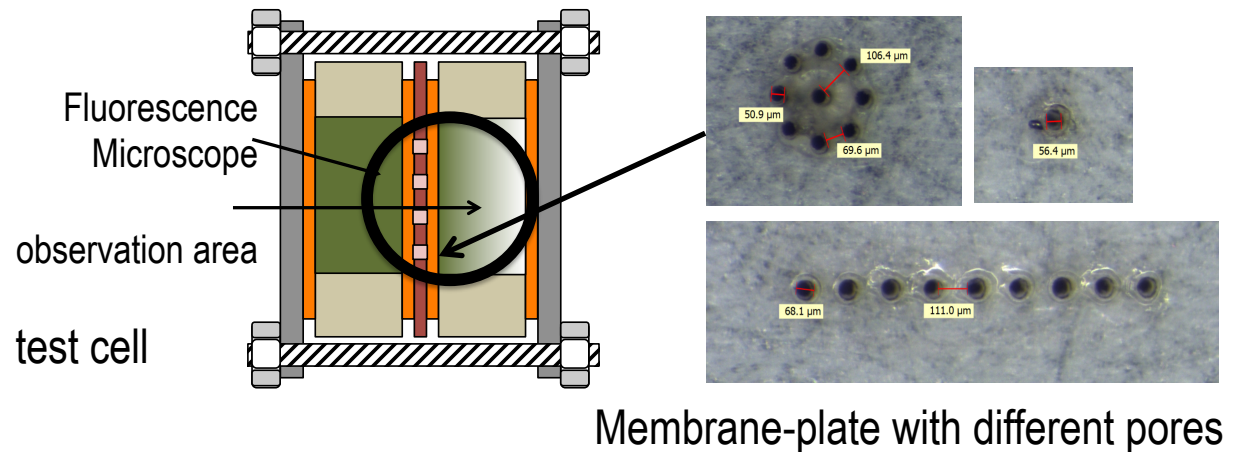
Controlled release of growth factors (GF)* from a localized well

*e.g. such as brain-derived neurotrophic factor (BDNF)

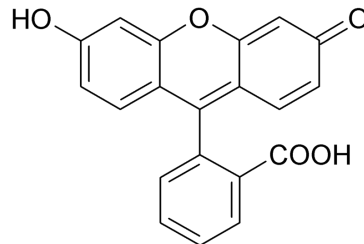
Objective:



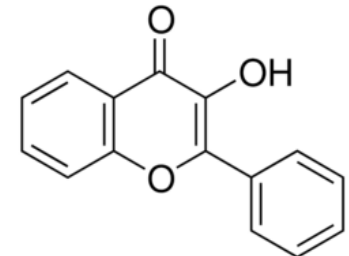
Experimental:



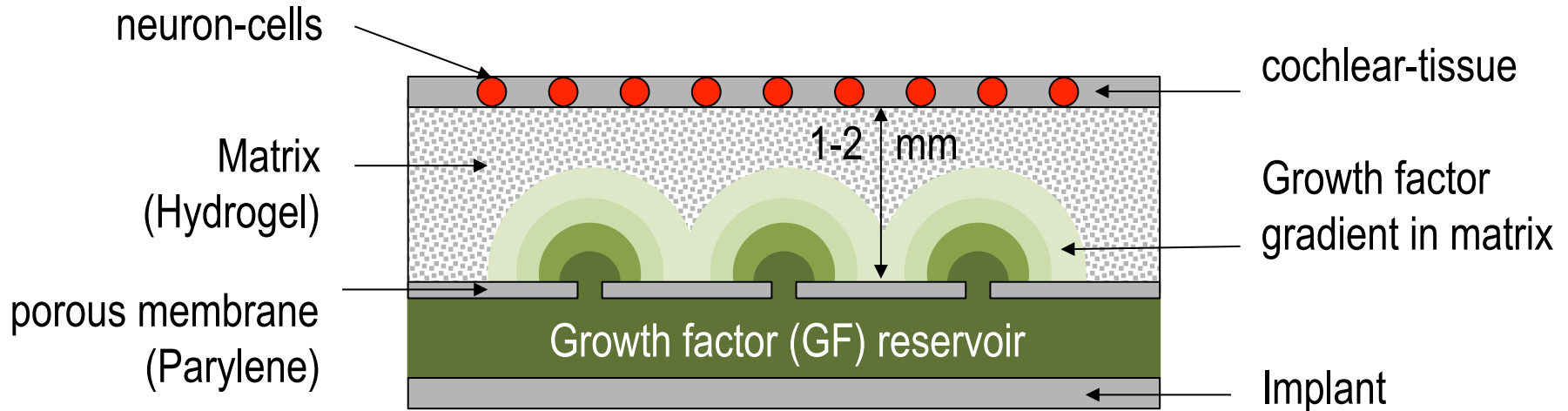
GF monitoring:



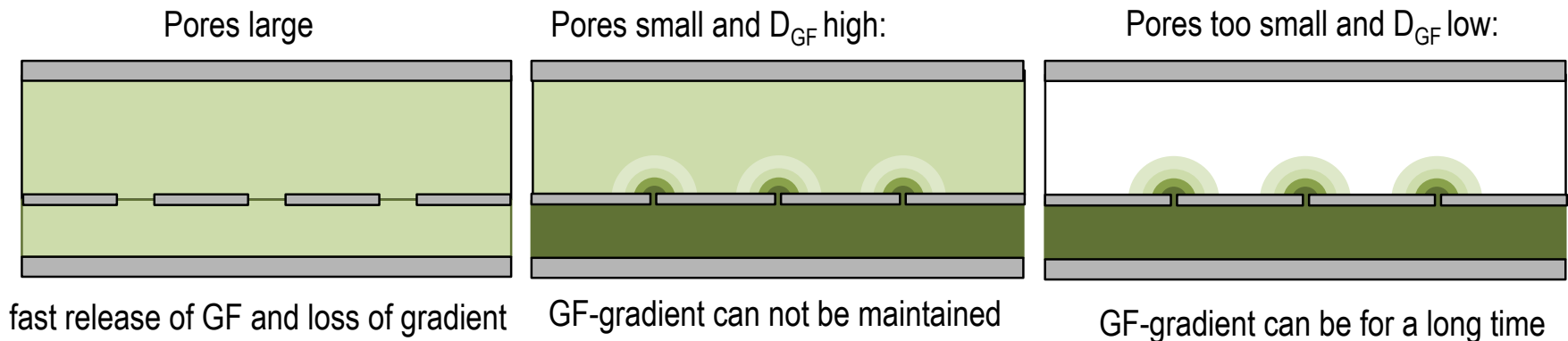
← Fluorescein:
marker + similarity
to Dihydroxyflavone
(possible GF) →



Scenarios for pore-size optimization



Cases study

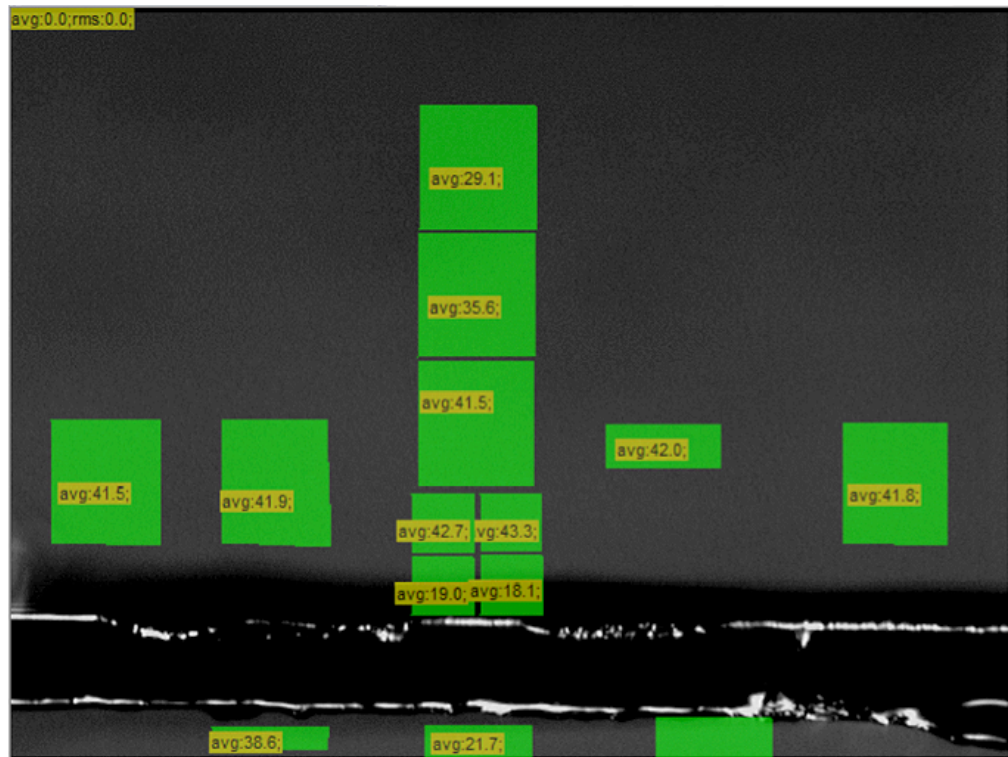


Design requirements:

The gradient of growth factors (GF) in the matrix (M) must be maintained during **4 days across 1-2 mm** in order to allow oriented grow of the neuron cells toward the implant prior to be attached with the contact-pad.

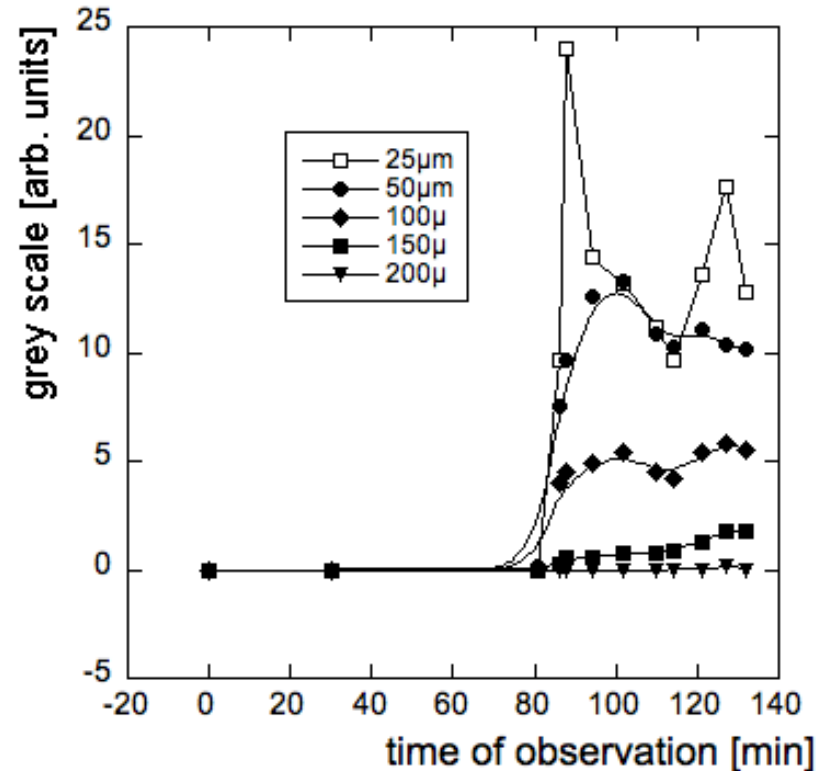
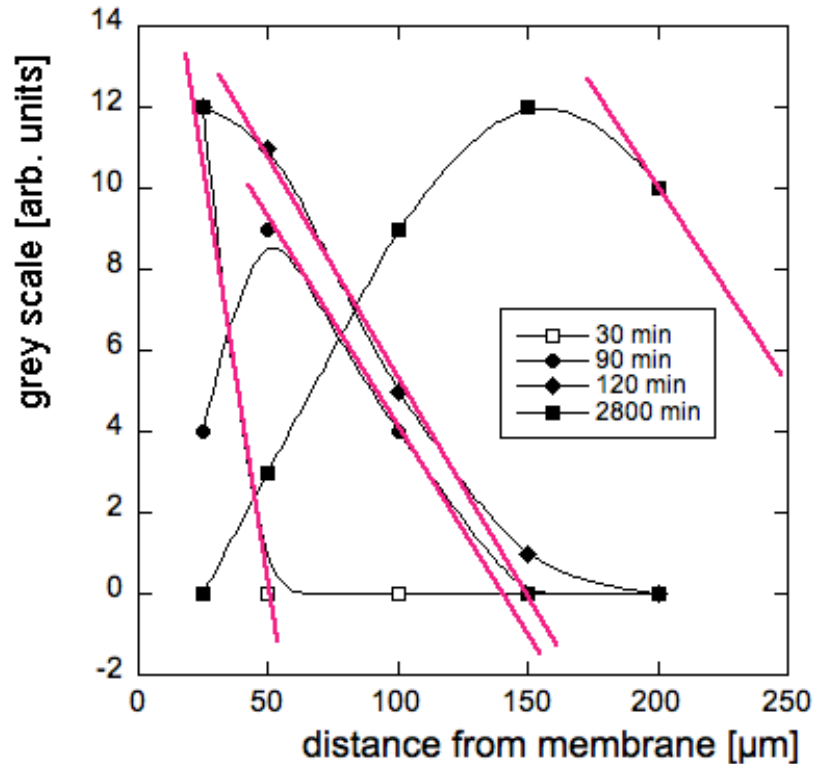
After the choice of M and GF the diffusion constant $D_{GF}(M)$ and the mobility $\mu_{AGF}(M)$ of GF in M will be determined (dependence on the molecular size of GF, the **pore size**, and **diffusion coefficient of GF** in M).

Fluorescence microscope
Micrograph observing the
spatially resolved intensity
of light emitted by the
Fluorescein molecules



Results

A 25 mmol/l fluorescein in glycerol solution diffuses through a single 50 μm opening

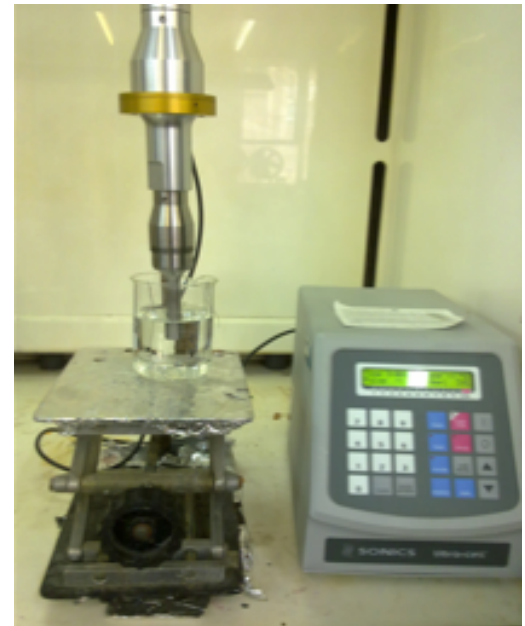
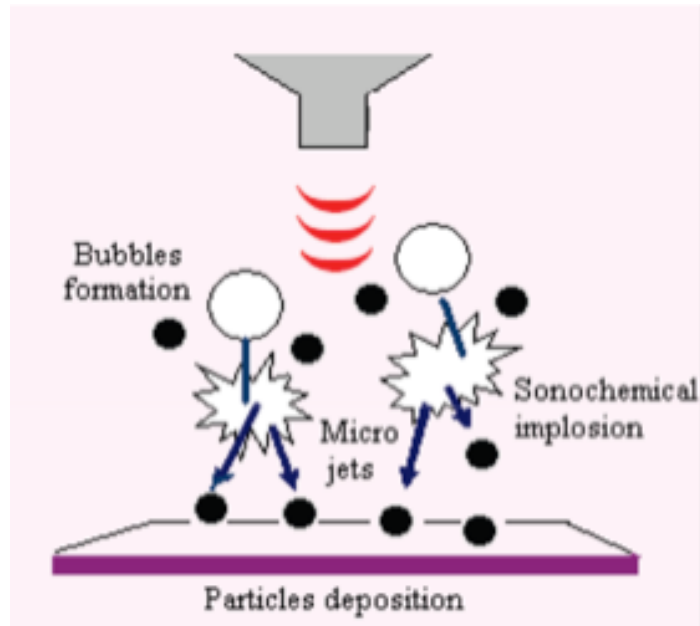


With a diffusion constant of $1.48 \cdot 10^{-5} \text{ mol / l} \cdot \text{s} \cdot \text{m}$ and creates a gradient of concentration of $2.75 \text{ mol / l} \cdot \text{m}$

Conclusion: the observed system could be potentially be used if fluorescein would be replaced by DHF

2) Deposition of nanoparticles by sonochemistry

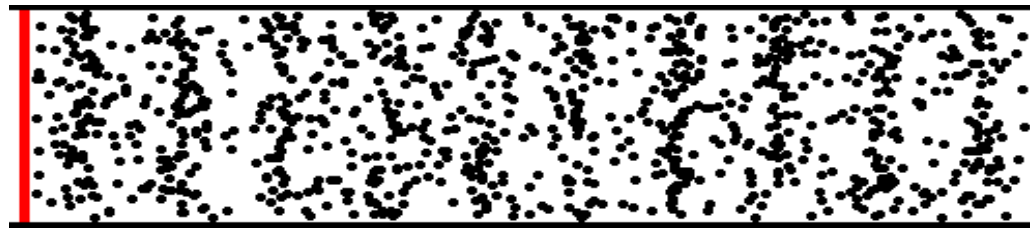
Pioneer input from A. Gedanken / Bar ilan University



Nanoparticles in liquids are interacting with a high frequency acoustic field (20 kHz 3 kW)

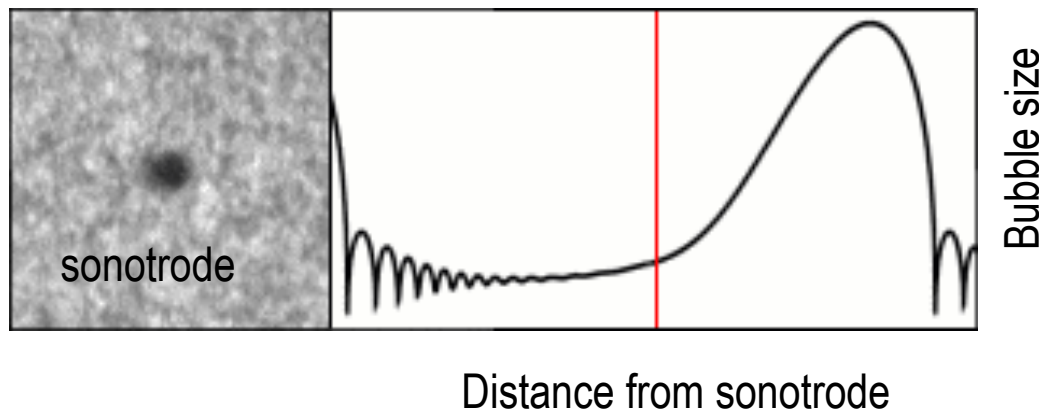
principle of sonochemistry

Sound = pressure waves = periodic compression / expansion- cycles traveling through a medium possessing elastic properties (gas, liquid, solid)

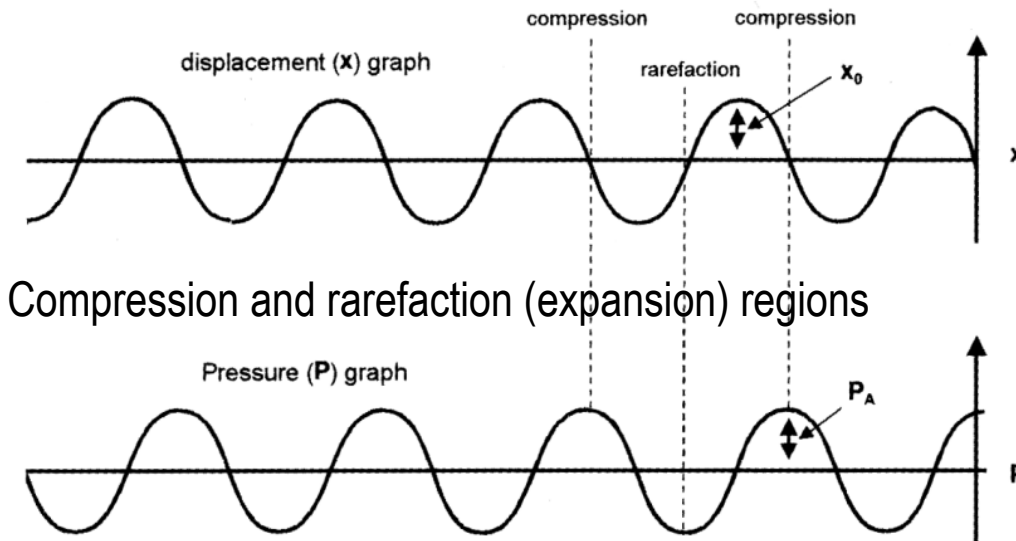


20 kHz

Acoustic Cavitation = creation, growth, and implosive collapse of bubbles in a liquid



Further considerations: Acoustic pressure



Compression and rarefaction (expansion) regions

$$P_A = \sqrt{2I\rho c}$$

P_A = driving pressure amplitude [Pa]; I = irradiation intensity [W m^{-2}]

(500 W system - $1.3 \cdot 10^5 \text{ W m}^{-2}$); ρ = liquid density [kg m^{-3}]

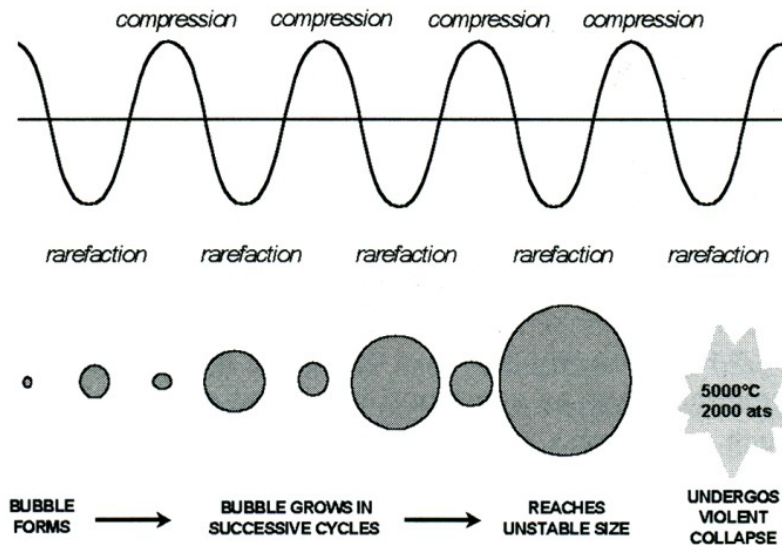
c = sound velocity in liquid [m s^{-1}] (Water 1482 m s^{-1})

$P_A = 620\,700 \text{ Pa} = 6.2 \text{ bar}$

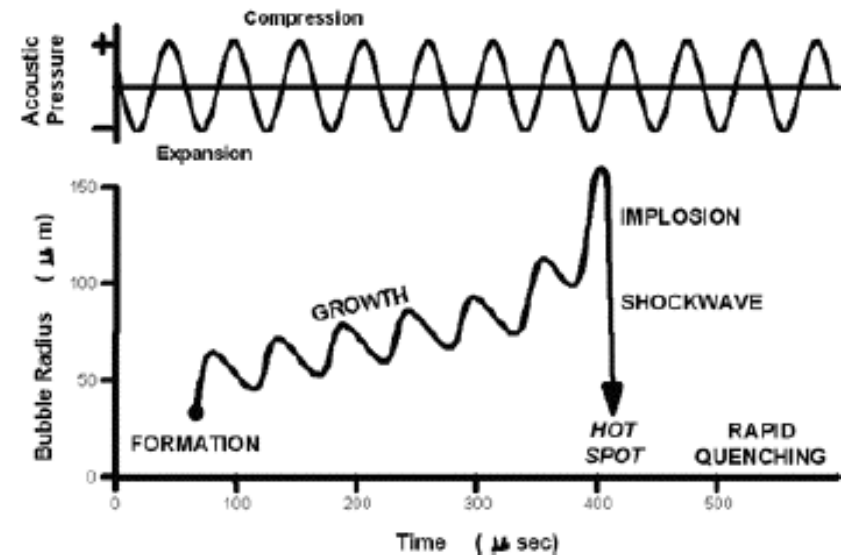
Acoustic Cavitation

Bubbles collapse = spherically symmetrical implosion,
shear forces, adiabatic compression, life time 1-2 μs

Hot spot = end of the collapse temperature of the gas inside bubble 5000–20 000°C (for 1 ns)
surrounding liquid layer 2000 °C pressure 500 –1500 bar Extreme cooling rates 10^{10}Ks^{-1}



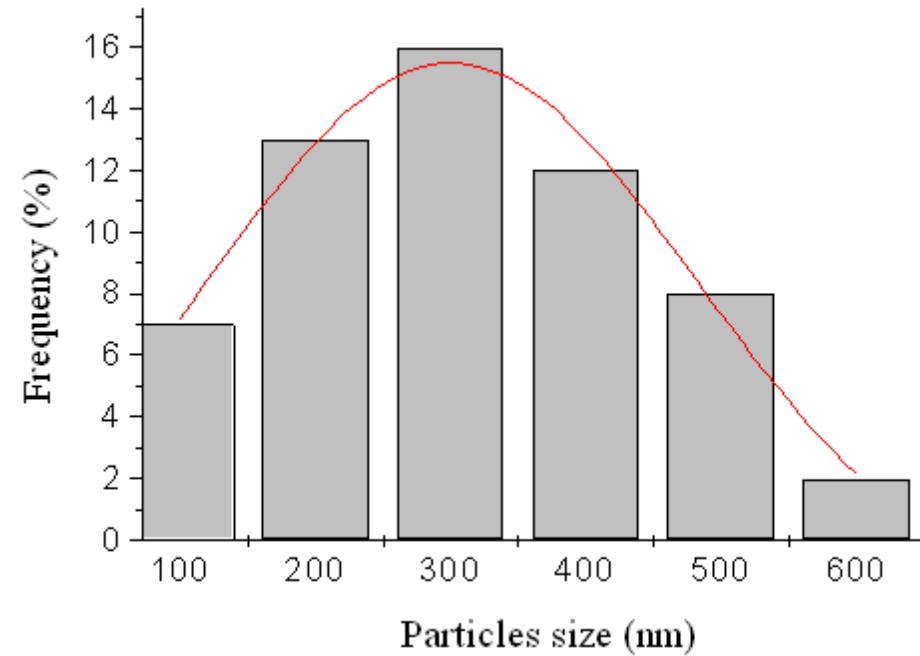
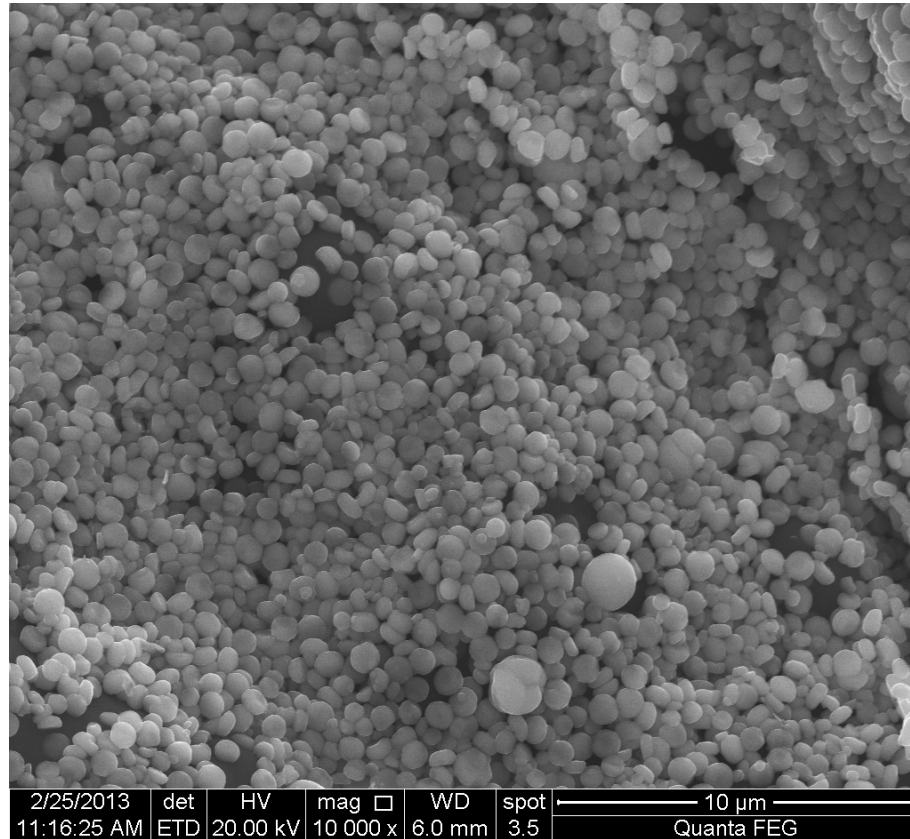
TRANSIENT CAVITATION: THE ORIGIN OF SONOCHEMISTRY



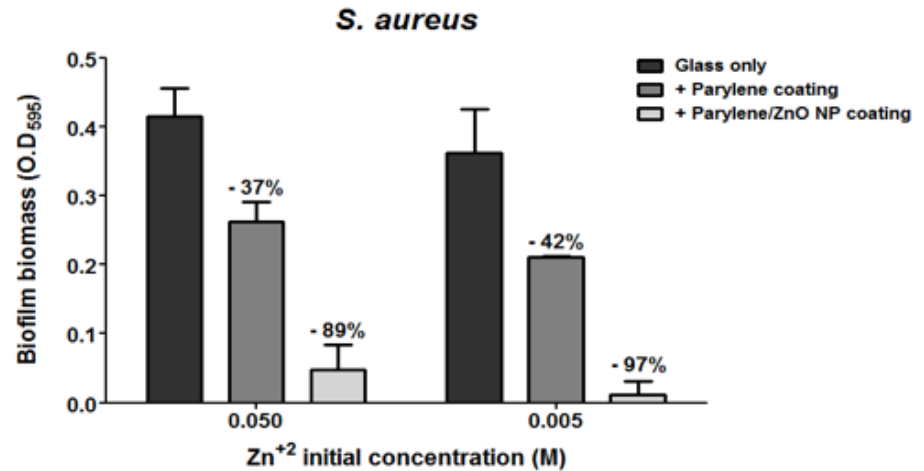
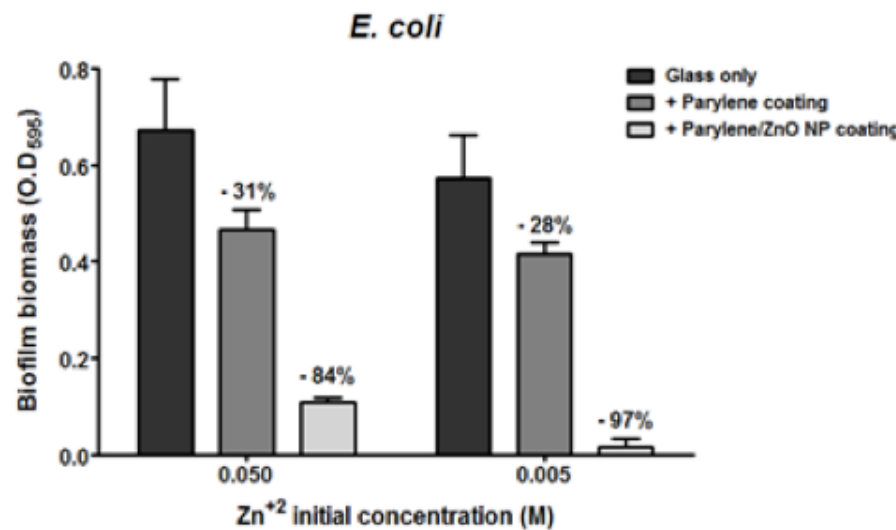
Summary: Sonochemical reactions

- Chemical changes / reactions induced by ultrasound
- No direct interaction of ultrasound fields with molecules
(in contrast to photons, heat, electro-magnetic fields)
- Liquid phase reactions driven by cavitation effects, radical generation, bond- breaking
- Solid state reactions - introduction of defects
- **Extreme acceleration of nanoparticles in the imploding bubble (shockwave-induced nano-particle implantation)**
- **Isotropic implantation in complex objects from all sides; hence cylindrical objects and tissues can be homogeneously be treated.**

Implantation of ZnO, MgF₂ nanoparticles using sonochemistry



Results



Biofilm biomass formation resulted after 24 hours of treatment with pure glass slides, glass slides coated solely with Parylene and glass slides coated with Parylene and ZnO nanoparticles. No leaching of nanoparticles was observed at long term.

Gedanken, A. Using Sonochemistry for the Fabrication of Nanomaterials. *Ultrason. Sonochem.* 2004, 11, 47-55.

Summary and Conclusion

Solid on liquid deposition is a platform technology allowing integrating liquids as functional “layer” in microsystems

Sonochemistry is a method allowing isotropic implantation of nanoparticles into all kind of surfaces such as metals, ceramics and even textiles.

Acknowledgement

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